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ORIGINAL ARTICLE

Application of 10% Ascorbic Acid Improves Resin Shear Bond Stregth in Bleached Dentin

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ABSTRACT

Restoration of the teeth immediately after bleaching with H_2O_2 35% is contraindicated due to the remnants of free radical that will stay inside dentin for 2-3 weeks which will compromise the adhesiveness of composite resin. **Objective:** The aim of this study was to evaluate the influence of 10% ascorbic acid on shear bond strength of composite placed on bleached dentin. **Methods:** Twenty seven samples were divided equally into three groups. Group 1: dentin was etched with 35% phosphoric acid; Group 2: dentin was bleached with 35% H_2O_2 followed by etching with 35% phosphoric acid; Group 3: dentin was bleached with 35% H_2O_2 , followed by application of 10% ascorbic acid and etched with 35% phosphoric acid. All samples were then stored at 37°C for 24 hours. The Universal Testing Machine was used to measure shear bond strength and the results were analyzed with Kruskal Wallis and Mann Whitney test. **Results:** After nine independent experiments, 10% ascorbic acid application on bleached dentin resulted in highest increased in bond strength (56.04±11.06MPa) compared to Group 2 (29.09±7.63MPa) and Group 1 (25.55±2.22MPa) and the difference was statistically significant (p<0.05). **Conclusion:** Application of 10% ascorbic acid to the bleached dentin improved the shear bond strength of resin composite.

ABSTRAK

Aplikasi asam askorbat 10% meningkatkan *shear bond strength* resin pada dentin pasca pemutihan gigi. Gigi pasca pemutihan gigi interna merupakan kontra indikasi untuk segera dilakukan restorasi, karena adanya radikal bebas yang masih tertinggal di dalam dentin selama 2-3 minggu dan dapat menggangu adaptasi bahan restorasi resin komposit. **Tujuan:** Menganalisis pengaruh aplikasi asam askorbat 10% terhadap *shear bond strength* resin komposit pada dentin pasca pemutihan gigi interna dengan gel H₂O₂ 35%. **Metode:** Dua puluh tujuh sampel gigi dibagi menjadi tiga kelompok. Grup 1: Dentin dietsa dengan asam fosfat 35%; Grup 2: Dentin dilakukan pemutihan gigi interna dengan gel H₂O₂ 35%, dietsa dengan asam fosfat 35%; Grup 3: Dentin dilakukan pemutihan gigi interna dengan gel H₂O₂ 35%, aplikasi asam askorbat 10% selama 10 menit, dan dietsa dengan asam fosfat 35%. Semua gigi percobaan di inkubasi pada 37°C selama 24 jam. Uji *shear bond strength* dilakukan dengan alat *Universal Testing Machine*. Data yang diperoleh dianalisis secara statistik dengan tes Kruskal Wallis dan Mann Whitney. **Hasil:** Analisis sembilan sampel dari masing-masing grup menunjukkan bahwa peningkatan *shear bond strength* yang paling banyak terjadi pada Grup 3 (56,04±11,06MPa) dibanding dengan Grup 2 (29,09±7,63MPa) dan Grup 1 (25,55±2,22MPa) dan perbedaannya bermakna secara statistik (*p*<0,05). **Simpulan**: Aplikasi asam askorbat 10% pada dentin pasca pemutihan gigi interna dengan gel H₂O₂ 35% dapat meningkatkan *shear bond strength* resin komposit.

Key words: 10% ascorbic acid, dental bleaching, dentin, shear bond strength

INTRODUCTION

Dentin is a calcified tissue and consists of tiny tubules or tubes. It is the second layer of a tooth and normally covered by enamel. Dentin covers the pulp tissue and constitutes the largest portion the tooth's structure. Dentin is denser and harder than bone and it's color may range anywhere from grey to black but typically is pale yellow. This yellow hue is generally what is seen penetrating through the tooth's enamel. Adhesion is a complex physical and chemical mechanism that allows the attachment of one substance to another. Good adhesion to dentin is one of the ideal properties of resin cement. Shear bond strength testing is one of the best measures of the adhesion.¹

Hydrogen peroxide (H_2O_2) is an active ingredient of dental bleaching agent. Thirty five percent of H_2O_2 has the highest bleaching effect and is available in gel form which is the safest formulation.² It is a strong oxidizer through the formation of free radicals.³⁻⁵ Free radicals are electrons which do not have partner, unstable, and highly reactive.⁵⁻⁹ Free radicals or reactive molecules produced by H_2O_2 will break the long-chained chromophore molecules, the cause of tooth discoloration, into smaller molecules with brighter colors.^{4,5}

After bleaching process, the free radical remnants will stay inside the tooth especially on dentin within 2-3 weeks. 10-12 These free radicals will compromise the adhesivity of composite resin because will distract the polymerization of composite resin due to a premature termination.^{13,14} It inhibits the binding process of the tooth structure and composite resin. Although, there was not much investigation on the bond strength of composite resin to the bleached dentin, almost all research, confirmed poor bond strength following bleaching treatment. It is believed that bond strength of composite resin to dentin is related to composite resin bond strength to enamel after bleaching process.¹¹ Almost all literatures which used 25-35% H₂O₂ showed neither shear bond strength nor tensile bond strength decreased on all composite resin type which are filled directly after bleaching procedure.

Antioxidant is a substance which has ability to neutralize free radicals after bleaching process. Ascorbic acid (vitamin C) is an antioxidant and it will react with superoxyde ($\rm O_2 \times$), hydrogen peroxyde ($\rm H_2 O_2$), hydroxyl ion (OH×). ^{14,15} The purpose of this study was to evaluate the effect of 10% ascorbic acid application to the shear bond strength of composite resin after bleaching with 35% $\rm H_2 O_2$.

METHODS

Twenty seven dentin samples from maxillary premolars were soaked in saline solution until use. Tooth crowns were sliced 2mm horizontally below the cervical area,

then cut into two pieces in bucco-palatal direction. Those halves part were fixed on 20mm diameters resin acrylic mould with dentin surface faced up and free from resin acrylic. The sample surfaces were ground flat using sand paper grit 800, 1200, 2000 and were polished using alumina paste until the surface was smooth and homogen.

The samples were divided equally into 3 groups. In Group 1, dentin surfaces were etched with 35% phosphoric acid (Scotchbond-Etchant, 3M ESPE) for 15 seconds, rinsed with water and dried. Bonding agent was then applied (Adper Single Bond 2, 3M ESPE) and cured for 10 seconds. Subsequently, nanohybrid composite resin (Filtek Z350 3M ESPE), 2mm in diameter and 3mm in thickness were applied to the dentin surfaces dan cured. In Group 2, 35% H₂O₂ (Opalescence Endo Ultradent USA) were applied to dentin surfaces, covered with plastic, placed on sterile plastic box and kept in the 37°C incubator for 3 days. After rinsed in running water and dried, the same procedures as Group 1 were applied. In Group 3, after the application of 35% H₂O₃ for 3 days, dentin surfaces were rinsed with water and dried. Ten percent of ascorbic acid (Harumsari Jakarta) were applied for 10 minutes and followed by the same procedures as Group 1. All samples were kept in the 37°C incubator for 24 h. Shear bond strength test for each samples was done using Universal Testing Machine with punch method.17,18

RESULTS

After nine independent experiments, 10% ascorbic acid application on bleached dentin resulted in highest increased in bond stregth (56.04±11.06MPa) compared to Group 2 (29.09±7.63MPa) and Group 1 (25.55±2.22MPa) and the difference was statistically significant (p<0.05). The range of shear bond strength value of each group was described in Table 1. The result showed that 10% ascorbic acid for 10 minutes can increase shear bond strength of composite resin filling to dentin surface post internal bleaching using 35% $\rm H_2O_2$ gel approximately two times of that without ascorbic acid.

Table 1. Measurement of shear bond strength of each experimental group using universal testing machine

| Experimental Group | n | Range Value (MPa) | Mean ± SD (MPa) | p |
|--------------------|---|----------------------|--------------------|--------|
| 1 | 9 | 23.23 - 30.25 | 25.55 ± 2.22 | 0.001* |
| 2 | 9 | 10.91 - 36.48 | 29.09 ± 7.63 | |
| 3 | 9 | 41.16 - 73.28 | 56.04 ± 11.06 | |

^{*} Kruskal Wallis, p<0,005.

Mann-Whitney test was conducted subsequently to find out the difference among each group of experiment. The result that each group had significant difference (p<0.005). With that result, the hypothesis of this study

was accepted, that the application of 10% ascorbic acid for 10 minutes increased shear bond strength of composite resin filling to dentin surface post internal bleaching using 35% H₂O₂ gel.

DISCUSSION

Thirty five percent H₂O₂ is a strong oxidizer and commonly used as dental bleaching agent. It produces very reactive free radicals, making this substance a very powerful oxidizing agent. Ascorbic acid, a free radicals reducing agent, was reported to reverse the negative effects of the free radicals. It will restore the altered redox potential of free radicals and allow polymerization of the resin without premature termination and thus reversing the compromised bonding.¹⁹ In the present study, 10% ascorbic acid was used to reduce the free radicals on dentin treated with 35% H₂O₂. According to Kimyai et al. (2008) 10% ascorbic acid is equally effective as the concentration of 20%.²⁰ It was also stated that low concentration of ascorbic acid could react with hydroxyl ion into ascorbil which was less reactive while high concentration do not react.15 It needs 10 minutes contact of ascorbic acid on tooth surface after bleaching with natrium hypoclorite and RC-Prep to remove the free radicals effect.²⁰

Our study confirmed that application of 10% ascorbic acid significantly improved shear bond strength is in accordance with the study conducted previous research.12 There was a significant difference of shear bond strength among the three groups. Group 3 showed the highest shear bond. It is hypothesized that the difference in bond strength is realted to the free radicals ability of 35% H₂O₂ that was quickly and completely absorbed.²² Dentin collagen also played an important role in increasing the cross-linking and stabilization of shear bond strength. Collagen in biological tissue is strengthened by the formation of native cross-links which provided the fibrillar resistance against enzymatic degradation as well as greater tensile properties. Inter molecular cross-linking is the prerequisite for stability, tensile strength and viscoelasticity of the collagen fibrils. The strengthening of collagen fibrils by cross-linking agents by increasing the mechanical properties and decreasing enzymatic degradation is a significant step in adhesive dentistry.²³

The lowest value of shear bond strength was when dentin only being etched by 35% phosphoric acid (Group 1) and followed by bleached with 35% $\rm H_2O_2$ and application of 35% phosphoric acid (Group 2). This showed that the application of 35% $\rm H_2O_2$ increased shear bond strength between dentin and composite resin. Even though the remnants of free radicals after bleaching process were still exist, it will compete with free radicals produced from polymerization of adhesive material. This further causes premature termination

resulted in incomplete polymerization of the adhesive material.^{13,14} Low value of shear bond strength on Group 1 was possibly caused by the retention made from etching and bonding process on dentin. To achieve good bond strength on dentin, collagen is needed to make a hybrid layer of dentin to composite resin. Due to the *in vitro* nature of the study, the dentin samples were dehydrated and was lack of collagen.

CONCLUSION

The application of 10% ascorbic acid could improve the shear bond strength significantly in dentin treated with 35% H₂O₂.

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