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Hadi Kuncoro

1. Laboratory of Pharmaceuticals Research and Development, Tropical Pharmaca, Faculty of Pharmacy, UniversitasMulawarman, Samarinda 75119, Indonesia. 2. Departement of Chemistry, Faculty of Mathematics and Natural Sciences, Universitas Padjadjaran, Jatinangor 45363, Indonesia

Kindi Farabi

Departement of Chemistry, Faculty of Mathematics and Natural Sciences, Universitas Padjadjaran, Jatinangor 45363, Indonesia

Laode Rijai Laboratory of Pharmaceuticals Research and Development, Tropical Pharmaca, Faculty of Pharmacy, UniversitasMulawarman, Samarinda 75119,Indonesia

Euis Julaeha Departement of Chemistry, Faculty of Mathematics and Natural Sciences, Universitas Padjadjaran, Jatinangor 45363,Indonesia

Yoshihito Shiono Departement of Bioresources Engineering, Faculty of Agriculture, Yamagata University, Tsuruoka-shi, Yamagata 997-8555, Japan

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Cover Page Footnote

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Authors

Hadi Kuncoro, Kindi Farabi, Laode Rijai, Euis Julaeha, Yoshihito Shiono, and Unang Supratman

A Known Naphthalene, Isoeleutherol, from the Herb of Lygodium microphyllum

Hadi Kuncoro^{1,2}, Kindi Farabi², Laode Rijai¹, Euis Julaeha², Yoshihito Shiono³, and Unang Supratman^{2,4*}

1. Laboratory of Pharmaceuticals Research and Development, Tropical Pharmaca, Faculty of Pharmacy, Universitas Mulawarman, Samarinda 75119, Indonesia

2. Departement of Chemistry, Faculty of Mathematics and Natural Sciences, Universitas Padjadjaran, Jatinangor 45363,

Indonesia

3. Departement of Bioresources Engineering, Faculty of Agriculture, Yamagata University, Tsuruoka-shi, Yamagata 997-8555, Japan

4. Central Laboratory of Universitas Padjadjaran, Jatinangor 45363, Indonesia

*E-mail: unang.supratman@unpad.ac.id

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Abstract

A known naphthalene, isoeleutherol (1), was isolated from the herb of *Lygodium microphyllum*. The chemical structure of 1 was determined on the basis of spectroscopic data mainly UV, IR, HRTOFMS, 1D- and 2D-NMR spectroscopy, as well as by comparing with compounds previously reported. Isoeleutherol was isolated from this plant for the first time and showed moderate antioxidant activity against DPPH (2,2-diphenyl-1-picrylhydrazyl) with IC₅₀ value of 53.96 \pm 2.87 µg/mL.

Abstrak

Senyawa Naftalena yang telah dikenal, Isoeleutherol, dari Herba *Lygodium microphyllum*. Senyawa naftalena yang telah dikenal, isoeleutherol (1) telah diisolasi dari herba *Lygodium microphyllum*. Struktur kimia senyawa 1 telah ditentukan berdasarkan data spektroskopi terutama UV, IR, HRTOFMS, NMR 1D- dan 2D-serta perbandingan dengan senyawa yang mirip dari laporan sebelumnya. Isoeleutherol diisolasi dari tanaman ini untuk pertama kalinya dan menunjukkan aktivitas antioksidan yang sedang terhadap DPPH (2,2-difenil-1-pikrilhidrazil) dengan nilai IC₅₀ 53.96 ± 2.87 µg/mL.

Keywords: Naftalena, Lygodium microphyllum, isoeleutherol, Lygodeceae, antioxidant activity

Introduction

Lygodium is only one genus of the Lygodiaceae family consisting of more than 50 species, and has been reported for the treatment of kidney stones [1] and as an expectorant [2], as well as for its antiplasmodial [3], antibacterial [4], antiviral [5], and antidiarrheal activities [6]. Phytochemical studies on the *Lygodium* have reported it to contain unique secondary metabolites with diverse biological activities. These metabolites include flavonoids [7-9], phenolic glycosides [10], naphthoquinones [11], ecdysteroids [12], and steroids [13]. As part of our studies on the Indonesian *Lygodium* species, we have performed a phytochemical examination of the herb of *L. microphyllum*.

The plant, known as "krokot" in Indonesia, is a perennial fern that typically grows in the rain forest and can be found on Kalimantan island [14]. This plant is used in Indonesia folk medicine for the treatment of fever and kidney stones [1-3]. In previous papers, we reported the isolation of flavonoids from the herb of *L. microphyllum* [8-9]. In this paper, we present the isolation of a known naphthalene derivative, isoeleutherol, and its antioxidant activity.

Materials and Methods

General. The melting point was measured on electrothermal melting point apparatus and not corrected. The

UV-Visible spectrum was obtained on a Shimadzu series 1800 spectrophotometer (Shimadzu, Kyoto, Japan). The IR spectrum was measured on a Perkin-Elmer 1760X spectrophotometer (Waltham, MA, USA) in KBr. Mass spectra recorded with a Waters, QToF HR-MS XEVotm mass spectrometer (Waters, Milford, MA, USA). ¹H and ¹³C NMR spectra are obtained with a JEOL NMR A-500 MHz using tetramethylsilane (TMS) as an internal standard (JEOL, Tokyo, Japan). Chromatographic separations were carried out on silica gel 60 (Merck, Darmstadt, Germany), ODS (Fuji Silysia, Kyoto, Japan). TLC plates were precoated with silica gel GF₂₅₄ and RP-18 (Merck, 0.25 mm) and detection was achieved by spraying with 10% H₂SO₄ in ethanol, followed by heating and under ultraviolet-visible light at wavelength of 257 and 364 nm. Preparative MPLC using a Buchi Pump Controller C-610, Buchi Pump Modules C-605, and a FLH-R10030B SiliCycle column- ISO04 SiliasepTM (Buchi, Swizerland).

Plant material. The herb of *L. microphyllum* was collected from forest areas in Samarinda, East Kalimantan in June 2016. The plant was identified by staff at the Faculty of Forestry, University of Mulawarman, Samarinda and sample specimens (No. 02042013) were stored at the Faculty of Forestry, University of Mulawarman, Samarinda, Indonesia.

Extraction and Isolation. The dried herbs (2.5 kg) of L. microphyllum were extracted with methanol (12 L) at room temperature for 4 days. After removal of the solvent under vacuum, the viscous concentrated MeOH extract (210 g) was suspended in H₂O and partitioned with n-hexane, EtOAc, and n-butanol, successively. Evaporation of the solvents resulted in n-hexane (59 g), EtOAc (72 g), and n-butanol (54 g) extracts. The nhexane soluble fraction (30 g) was fractionated by vacuum liquid chromatography on silica gel 60 using a gradient of n-hexane and EtOAc to give ten fractions (A-J). Fraction C (3.4 g) was chromatographed on a column of silica gel, using a gradient of n-hexane-EtOAc (10:0-1:1), to give eight fractions (C1-C8). Subfraction C6 (450 mg) was separated by using MPLC on silica gel, eluted with CHCl₃-EtOAc (9:1) to give five fractions (C6.1-C6.5). Subfraction C6.3 (86 mg) was separated on preparative TLC, eluted with nhexane:EtOAc:HOAc = 9:1:0.5, R_f value of 0.42, to give 1 as a brown crystals. Compound 1 had an R_f value of 0.45 on silica gel using n-hexane:EtOAc (7:3) and 0.38 on silica RP-18 using MeOH: H_2O (3:2).

Antioxidant assay. The DPPH (2,2-diphenyl-1-picryl-hydrazyl) method was used to evaluate radical scavenging activity. A 10 μ L aliquot of each extract sample was added to 990 μ L of DPPH solution (0.002% in methanol). The mixture was incubated for 30 minutes at room temperature, the absorbance was measured at

517 nm against a corresponding blank, and the antioxidant activity was calculated as: AA% = ($A_{DPPH} - A_{sample}$) / $A_{DPPH} \times 100$

where AA is the antioxidant activity, A_{DPPH} is the absorption of DPPH against the blank, and A_{sample} is the absorption of the extract or control against the blank. Ascorbic acid was used as positive control. All tests were carried out in triplicate [15].

Results and Discussion

Isoeleutherol (1) was isolated as brown crystals, m.p. 203-204 °C, optical rotation $[\alpha]^{20}_{D}$ -64° (*c* 0.10, CHCl₃) (Figure 1). The molecular formula of compound 1 is designated as C14H12O4 based on the HRTOF-MS spectrum $(m/z \ 245.0817 \ [M-H]^+, \text{ calcd. for } C_{14}H_{13}O_4,$ m/z 245.0814) and NMR data (Table 1), thus requiring nine degrees of unsaturations. IR spectrum of 1 showed the presence of hydroxyl (3420 cm^{-1}), ester (1710 cm^{-1}), benzene (1610 and 780 cm^{-1}), and ether (1210 cm^{-1}) groups, while the UV spectrum showed absorption maxima at 320, 275, and 150 nm, indicating the presence of aromaticity in **1**. The ¹H-NMR (acetone- d_6) spectrum of 1 revealed the ABC-type signals at δ_H 7.70 (1H, d, J=8.5 Hz), 7.51 (1H, dd, J=8.5, 7.8 Hz), and 7.17 ppm (1H, d, J=7.8 Hz), which implied the presence of a trisubstituted benzene ring in 1. A resonance signal at $\delta_{\rm H}$ 7.88 ppm (1H, s) in the ¹H-NMR spectrum of **1** indicated the presence of a pentasubstituted benzene ring. The ¹H-NMR spectrum also showed a signal for an alkyl group at [$\delta_{\rm H}$ 1.69 (3H, d, J=6.5 Hz), 5.71 ppm (1H, q, J=6.5 Hz)], a methoxyl group at $\delta_{\rm H}$ 4.21 ppm (3H, s), and a hydroxyl proton at δ_H 9.88 ppm (1H, s). The ¹³C-NMR spectrum showed 14 carbon resonances, which were classified by their chemical shifts and DEPT spectra as one methyl, one methoxy, four sp^2 methines, six sp² quaternary carbons, one sp³ methine and one lactone carbon.

These functionalities accounted for six out of the total nine degrees of unsaturation. The remaining of three degrees of unsaturation were consistent with a naphthalene skeleton-[13]. The gross structure of **1** was deduced from the ¹H-¹H COSY and HMBC spectra (Figure 2). The hydroxyl proton at $\delta_{\rm H}$ 9.88 ppm was correlated to $\delta_{\rm C}$ 118.3 (C-12)

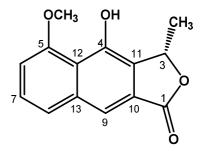


Figure 1. Structure of Isoeleutherol (1)

Position of C	1 (acetone-d ₆)		Eleuthero [14] (DMSO-d ₆)		Isoeleutherol (CDCl ₃) [15]	
	¹ H-NMR $\delta_{\rm H}$ (int., mult., <i>J</i> Hz)	13 C-NMR δ_{C} (mult.)	¹ H-NMR $\delta_{\rm H}$ (int., mult., <i>J</i> Hz)	13 C-NMR δ_{C} (mult.)	¹ H-NMR $\delta_{\rm H}$ (int., mult., <i>J</i> Hz)	13 C-NMR δ_{C} (mult.)
1	-	170.3 (s)	-	170.1 (s)	-	170.5 (s)
3	5.71 (1H, q, 6.5)	77.5 (d)	5.77 (1H, m)	77.2 (d)	5.70 (q, 6.5)	76.6 (d)
4	-	150.2 (s)	-	116.2 (s)	-	149.2 (s)
5	-	157.7 (s)	-	156.7 (s)	-	156.6 (s)
6	7.17 (1H, d, 8.5)	107.5 (d)	7.12 (1H, d, 7.8)	107.3 (d)	6.93 (1H, s, 7.8)	106.3 (d)
7	7.51 (1H, dd, 8.5, 7.8)	123.9 (d)	7.48 (1H, dd, 8.3, 7.8)	123.2 (d)	7.54 (1H, d, 8.3)	123.7 (d)
8	7.70 (1H, d, 7.8)	128.7 (d)	7.69 (1H, d, 8.3)	127.7 (d)	7.54 (1H, d, 8.3)	127.6 (d)
9	7.88 (1H, s)	116.6 (d)	7.93 (1H, s)	149.4 (d)	7.84 (1H, s)	116.5 (d)
10	-	126.8 (s)	-	125.5 (s)	-	125.9 (s)
11	-	128.0 (s)	-	128.1 (s)	-	127.9 (s)
12	-	118.3 (s)	-	117.5 (s)	-	117.5 (s)
13	-	138.3 (s)	-	137.3 (s)	-	137.2 (s)
3-CH ₃	1.69 (3H, d, 6.5)	19.5 (q)	1.64 (1H, d, 1.2)	19.4 (q)	1.73 (1H, d, 6.5)	19.2 (q)
5-OCH ₃	4.21 (3H, s)	57.1 (q)	4.04 (s)	56.4 (q)	4.11 (1H, s)	56.9 (q)
4-OH	9.88 (1H, s)	-	-	-	9.63 (1H, s)	-

Table 1. NMR for 1, Eleutherol [14] and Isoeleutherol [15]

and 128.0 ppm (C-11), and the aromatic proton at $\delta_{\rm H}$ 7.88 ppm was correlated to δ_C 126.8 (C-10) and 138.3 ppm (C-13), which indicated that the hydroxyl group and aromatic proton were located at C-4 and C-9, respectively. The methoxyl proton at $\delta_{\rm H}$ 4.21 ppm was correlated to $\delta_{\rm C}$ 157.7 ppm (C-5), which indicated that the methoxy group was located at C-5. The methyl proton at $\delta_{\rm H}$ 1.69 ppm was correlated to $\delta_{\rm C}$ 77.5 (C-3) and 128.0 ppm (C-11), which indicated that the methyl group was located at C-3. The signal of an oxygenated sp³ methine at δ_H 5.71 ppm was correlated to the carbonyl lactone at $\delta_{\rm C}$ 170.3 ppm (C-1), which indicated that the lactone ring was formed between C-10, C-1, and C-3. Aromatic signals at δ_H 7.17 and 7.70 ppm were correlated to δ_H 157.7 (C-5) and 138.3 ppm (C-13), whereas another aromatic signal at δ_H 7.51 ppm was correlated to δ_C 107.5 (C-6) and 128.7 ppm (C-8), which suggested the presence of ABC aromatic protons in 1.

In addition, the ¹H-¹H COSY spectrum showed a correlation of H-6, H-7, and H-8, which supported the presence of ABC aromatic protons from a trisubstituted benzene ring.

A detailed comparison of the NMR data of **1** to those of eleutherol and isoeleutherol [14,15], revealed that the structure of the compound **1** is more closely related to isoeleutherol rather than eutherol. The assignment as isoeleutherol was supported also by comparing the measured specific optical rotation of **1** $[\alpha]_{D}^{20}$ -64° (*c* 0.10, CHCl₃) to that of isoeleutherol ($[\alpha]_{D}^{18}$ -60.5° (*c* 0.5, CHCl₃)] [14] and eleutherol ($[\alpha]_{D}^{18}$ +83° (*c* 0.373, CHCl₃)] [15,16]. Consequently, compound **1** was identified as isoeleutherol.

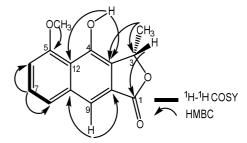


Figure 2. Selected ¹H-¹H COSY and HMBC Correlations for 1

Isoeleutherol (1) was evaluated for its radical scavenging activity using a DPPH (2,2-diphenyl-1-picrylhydrazyl) assay, using ascorbic acid as a positive control [17]. Isoeleutherol showed moderate antioxidant activity with an IC₅₀ value of $53.96 \pm 2.87 \,\mu\text{g/mL}$.

Conclusions

A known naphthalene, isoeleutherol (1) has been isolated from the herb of *Lygodium microphyllum*. Isoeleutherol 1 was isolated from this plant for the first time and showed moderate antioxidant activity in a DPPH (2,2diphenyl-1-picrylhydrazyl) assay. Because it was previously unknown that *Lygodium microphyllum* contained isoeleutherol, it may be of interest to test isoeleutherol for activity in a variety of additional assays, such antiplasmodial, antibacterial, and antiviral assays."

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