

Optimization of PVA-Arabic Gum–Honey-based Electrospun Nanofibers as Candidate Carrier for Peptide and Protein Delivery

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ABSTRACT

Nanofibers dressing exhibit several advantageous characteristics for accelerating wound healing, such as its similar structure to the extracellular matrix (ECM), high surface area/volume ratio, high porosity and high loading capacity of drug. The nanofibers dressing which were prepared by the electrospinning technique using combination of synthetic and natural polymer excipients capable of fulfilling the ideal wound dressing criteria. This study aimed to develop nanofibers dressing prepared from polyvinyl alcohol (PVA), Arabic gum (GA) and honey by electrospinning method. This study focused on the effect of electrospinning parameters and the morphology of electrospun nanofibers of the blended solutions made from PVA (9% w/v) – GA (1% w/v) and honey (at varied concentrations of 0;1;3% w/v) with or without Triton X-100 (0.05% v/v). The effect of varied process parameters such as voltage and flow rate in electrospinning was also investigated. The blended solutions with the various concentration of honey at 0;1;3% w/v and Triton X-100 were named as FAt, FBt, and FCt, respectively, while the blended solutions without Triton X-100, were named as FA, FB, and FC. The optimum electrospinning parameter were 18 KV and 5 μ l/minute for FAt, FBt, and FCt; and 20 KV and 10 μ l/minute for FA, FB, and FC. Electrospun nanofibers of FAt, FBt, FCt showed smoother and more uniform fibers in comparison to the nanofibers FA, FB, and FC. The average nanofibers diameter of FAt, FBt, FCt were 244 ± 45 ; 266 ± 45 ; 283 ± 57 nm, respectively, while the average nanofibers diameter of FA, FB, FC was 406 ± 140 , 457 ± 168 , 594 ± 204 nm, respectively. Higher concentration of honey increased the diameters of nanofibers. The average nanofibers diameter of FAt, FBt, and FCt were within nanoscale in range of the ECM (50-500 nm), which were suitable for accelerating wound healing. Therefore, this study indicated that PVA-GA-honey nanofibers dressing is suitable to be further developed as carrier for growth factors.

Keywords: arabic gum; electrospinning; honey; nanofibers; polyvinyl alcohol

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INTRODUCTION

Wound healing process is a very complex process consisting haemostasis, inflammation, proliferation and remodeling phases (Orue et al., 2017). All phases of the wound healing process are controlled by active endogenous compounds, one of them is growth factors. Growth factors are biologically active polypeptides that play a crucial role in regulating wound healing process in which involved in cell proliferation, differentiation and migration (Gainza et al., 2015). Currently, treatment with growth factors require high doses and is administered over long periods of time. The conventional formula cannot prevent the rapid degradation rate of growth factors (Park et al., 2017). Therefore many research have been performed to develop delivery system for growth factors. Among them is by loading growth factor into nanofibers wound dressing formula by electrospinning to protect the growth factor from in vivo degradation, to improve its stability and control its release (Orue et al., 2017).

Nowadays the fabrication of wound dressing formula has attained higher criteria based on the principle of creating and maintaining a moist wound environment,

in which accelerating wound healing, skin regeneration, oxygen exchange, and inhibiting bacterial infection (Felgueiras et al., 2017). Electrospun nanofibers considerable promising as wound dressing. It has several structural advantages for accelerating wound healing process due to their physical structure which mimics to the three-dimensional structure of natural extracellular matrix as well as their high ratio of surface area to volume and a high interconnected porosity (Zhang et al., 2017). Besides that, the nanoscale size of the fibers causes the nanofibers dressing able to function as a barrier for preventing bacterial contamination (Asli et al., 2016). In addition, nanofibers dressing also demonstrated a high drug-loading capacity and additionally, they provide an tremendous protection from environmental harms (Orue et al., 2017).

The selection of the suitable material for nanofibers dressing is a challenge. Various synthetic polymers, natural polymers and the mixtures of both have been successfully fabricated for the development of nanofibers dressing. Synthetic polymers provide many advantages as compared to natural polymers, due to its degradability and its ability to produce consistent batch-to-batch

characteristics including mechanical properties (elastic and rigid). Their slow rate of can promote controlled release of the loaded drugs. Synthetic polymers are also more convenient to spin because they are compatible with various solvents. However, synthetic polymer also have limitations, such as low biocompatibility and the existence of solvent residues. Low biocompatibility will result in slower cell attachment and proliferation rates as compared to the natural polymers (Asli et al., 2017).

Natural polymers are of great interest to researchers as compared to synthetic polymers due to its biocompatibility, cheaper price and non-toxic characteristic. However, natural polymers showed less favorable mechanical properties and their degradation rates are relatively rapid, which limit their use to fabricate nanofibrous wound dressing, unless they are combined with synthetic polymer (Orue et al., 2017; Zhang et al., 2017). One of natural polymers is arabic gum (GA). Previous studies reported that arabic gum solution could form a smooth and uniform nanofiber by mixing the solution with synthetic polymers such as polyvinyl alcohol (PVA) (Padil et al., 2016). In addition, Bhatnagar et al (2013) recently also reported that arabic gum has potential as wound dressing due to its antibacterial and hemostatic activity.

Nanofibrous dressing from combination of both synthetic and natural polymer offer characteristic which can meet the ideal wound dressing criteria. To date, loading growth factors into the matrix of polyvinyl alcohol (PVA)-arabic gum (GA) has not been studied yet. PVA is a synthetic polymer that is easily soluble in water, biocompatible, electrospinnable and possess strong mechanical properties (Zhang et al., 2017). PVA can also be used for sensitive macromolecules such as proteins because its electrospinning process does not require organic solvents that have the potential to denature proteins (Asli et al., 2016). Honey was also added into the composite polymer to improve the stability of growth factors during the electrospinning process. (Wong et al., 2013). In addition, honey can improve the wound healing process because of its antibacterial, anti-inflammatory and antioxidants activity (Oryan et al., 2016).

The selection of the water instead of organic solvents give challenge in electrospinning because of its high surface tension. Therefore, Triton X-100 was added as surfactant to the formulation to reduce its surface tension and to increase spinnability of electrospinning blended solution. Additionally, during the electrospinning process, all parameters such as solution properties (viscosity, surface tension, and conductivity), process parameters (flow rate, tension, needle size, distance between needle and collector) and ambient parameters (temperature and humidity) were controlled. This

present study was carried out to optimize fabrication of nanofibers dressing using varied composition of blended solution and process parameters such as voltage and flow rate, and to evaluate the morphology of obtained nanofibers dressing.

METHODS

Materials

Gum arabic *spray-dried* (Merck, Germany), PVA (BM = 145,000 hydrolysis degrees 98.0-98.8 mol%) (Merck, Germany), deionized water, buffer phosphate (PBS), Indonesian honey, and Triton X-100 (Sigma-Germany).

Instruments

The instruments used in this study were Electrospinning 601 (Nachriebe, Indonesia), Analytical scales (Sartorius, Germany), Rotational viscometer (Meyer, USA), Conductometer (Mettler, USA), Tensiometer (Du noy, Germany), Scanning electron microscopy (Hitachi, JEOL JSM-6510LV, Japan), Magnetic stirrer, syringes and glassware (Pyrex).

Preparation of PVA-GA-Honey-based Nanofibers Dressing without Triton X-100

The electrospinning solutions were prepared based on Formula FA, FB, FC and FD (Table 1) by dissolving 0.9 gr PVA with hot water (temperature $\pm 80^{\circ}\text{C}$) and stirring for 2 hours using a magnetic stirrer. GA (0, 1 gr) was added to PVA solution, and mixed until homogen. Varied concentrations (0; 1; 3; and 5% w/v) of honey was then added to the solution, and deionized water were added until the total volume of 10 ml.

Preparation of PVA-GA-Honey-based Nanofibers Dressing with Triton X-100

The electrospinning solutions were prepared based on Formula FAt, FBt, and FCt (Table 2) by dissolving 0.9 gr PVA with hot water (temperature $\pm 80^{\circ}\text{C}$) and stirring for 2 hours using a magnetic stirrer. GA (0.1 g) was added to PVA solution, and mixed until homogen. Varied concentrations (0; 1; and 3% w/v) of honey and Triton X-100 (0.05% v/v) were then added to the solutions. Finally, deionized water were added until the total volume of 10 ml.

Electrospinning procedure

PVA-GA-Honey solutions were loaded in 5 ml syringe with a diameter of 0.7 mm. Afterwards, the nanofibers dressing were fabricated with the condition of electrospinning as follows (see Figure 1): the electrical voltage used was 16-20 kV, the flow rate of the solution was 5-10 $\mu\text{l}/\text{minute}$, the distance between the needle tip and collector was 18 cm and the collector rotation speed was 300 rpm. The electrospinning process was carried

Table 1. Formula and physical characteristic of PVA-GA-honey-based nanofibers dressing

Formula	PVA (% w/v)	GA (% w/v)	Honey (% w/v)	Viscosity (mPa.s)	Surface tension (dyne/cm)	Conductivity (μ S/cm)
FA	9	1	0	573 ± 21	$62,4 \pm 0,10$	$665,3 \pm 1,5$
FB	9	1	1	627 ± 15	$61,8 \pm 0,15$	$727,0 \pm 1,0$
FC	9	1	3	667 ± 15	$61,5 \pm 0,15$	$743,3 \pm 2,1$
FD	9	1	5	710 ± 10	$61,3 \pm 0,15$	$796,0 \pm 1,7$

Table 2. Formulation of PVA-GA-honey-based nanofibers dressing

Formula	PVA (% w/v)	GA (% w/v)	Triton X-100 (% v/v)	Honey (% w/v)
FAt	9	1	0,05	0
FBt	9	1	0,05	1
FCt	9	1	0,05	3

out at room temperature with relative humidity of $\pm 35\%$. The nanofibers dressing were then dried at room temperature in the desiccator.

Characterization of Nanofibers Dressing

Scanning Electron Microscope (SEM)

The surface morphology and the size of the nanofibers dressing was characterized using SEM (JEOL JSM-6510LV). Prior to observation, the nanofibers were coated with gold in order to be electrically conductive. Images were obtained at excitation voltage of 15 kV. Nanofiber diameters were then calculated at 100 points from SEM images processed using ImageJ software version 1.52a (USA).

RESULTS AND DISCUSSION

Characteristics of Blended Solution of PVA-GA-Honey Based

The color of the FA solution was transparent while the color of the solutions of FB, FC, and FD was yellowish to brownish due to addition of honey. The physical properties of all solutions were then measured and the results is depicted in Table 1.

The effect of honey addition with varied concentration on the viscosity, surface tension, and conductivity of the solutions FA, FB, FC, and FD can also be seen in Figure 2. Based on data from Table 1 and Figure 2, it showed that the viscosity and conductivity increased with increasing concentration of honey. Meanwhile, the surface tension of the solution was lower as the concentration of honey increased, but the difference is insignificant. The FA solution has a viscosity of 573 ± 21 mPa.s, while the viscosity of the solution of FB, FC, and FD was respectively 627 ± 15 mPa.s; 667 ± 15

mPa.s; and 710 ± 10 mPa.s. The addition of honey into the solution increased the viscosity of the solution since honey is a viscous solution. Honey is a natural mixture that contains very high sugar levels, namely glucose and fructose (85-95%) and sucrose. It also contains small amounts of amino acids, enzymes, organic acids, vitamins, minerals, and polyphenols (Maleki et al., 2012; Wong et al., 2013).

GA as an anionic hydrocolloid increased the conductivity of 9% PVA solution when the two solutions are mixed. FA solution showed conductivity value of 665.3 ± 1.5 μ S / cm, while the conductivity of the solution FB, FC, and FD, were respectively, 727.0 ± 1.0 μ S / cm; 743.3 ± 2.1 μ S / cm; and 796.0 ± 1.7 μ S / cm. This is because of the amount of ions present in the solution. The higher concentration of honey, the greater the conductivity of the solution. These results are similar with previous studies conducted by Sarhan, Azzazy, and El-Sherbiny (2016) who reported increase in the conductivity of the mixture of PVA and chitosan solution as the concentration of honey increase.

Optimization of Electrospinning Process PVA-GA-Honey Based on Formula Variations.

After the results of measurements of viscosity, surface tension, and conductivity of the FA, FB, FC, and FD formulas were obtained, the blended solution was used for electrospinning process at applied voltage 20 KV; flowrate 1 μ l/minutes; with horizontal distance of 18 cm from the needle tip to the collector and relative humidity $\pm 35\%$. There were only 3 formulas (FA, FB, and FC) which successfully spun into nanofibers dressing, while the FD solution was clogged at the needle tip during electrospinning since the solution was too viscous. Higher viscous solution causes the droplets of the

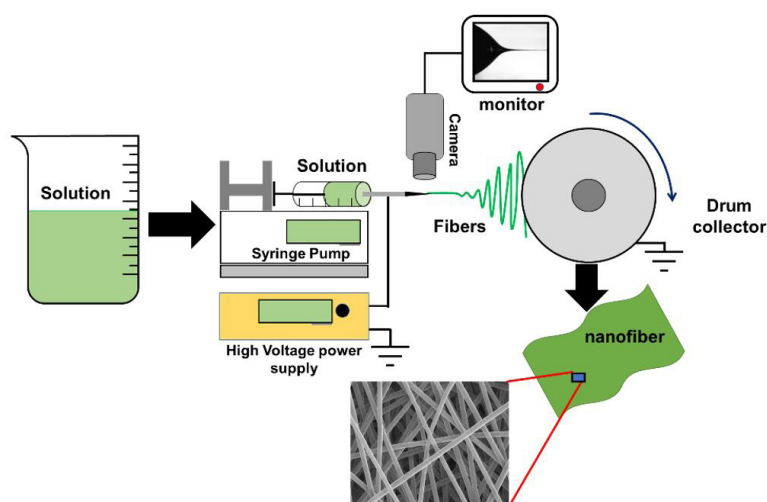


Figure 1: A diagram of the electrospinning system used for nanoscale production

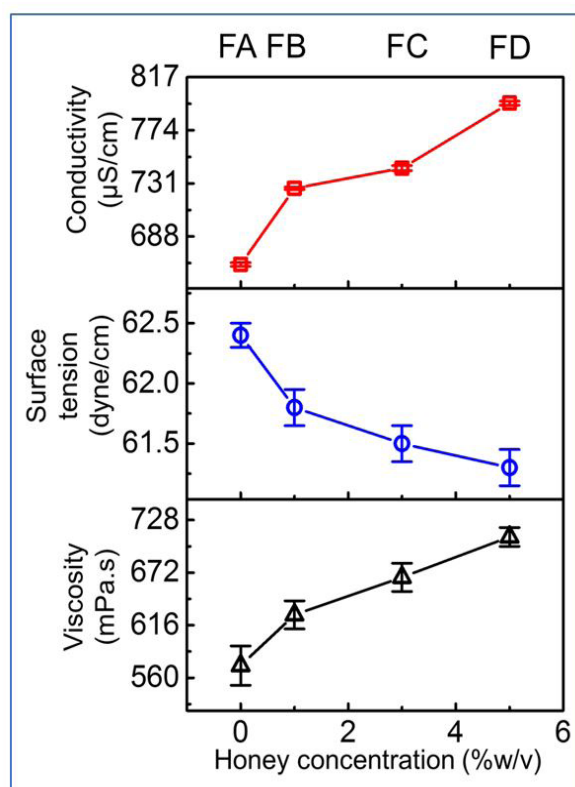


Figure 2. Effect of honey concentration on viscosity, surface tension and conductivity of the solution FA, FB, FC and, FD

solution on the tip of the needle to dry out easily during the electrospinning process because the amount of solvent to be evaporated is too small (Ramakrishna et al., 2005). Research conducted by Rezaei et al (2016) also reported the same thing, that the polymer mixture of almond gum and PVA with concentrations greater than 10% w/w was difficult to be spun since the solution gets clogged on the tip of the needle because of the solution was too viscous. Another study also stated that the optimum viscosity for the manufacture of PEO nano fibers is 800-4000

cP, while the optimal viscosity for the manufacture of polyacrylonitrile (PAN) polymeric nanoparticles is 1.7-215 cP (Haider et al., 2015).

The obtained nanofibers dressing were then morphologically characterized using SEM. The morphology and diameter size of the nanofibers was analysed using ImageJ software by selecting 100 fibers at random. The distribution curve and nanoscale diameter graph are created using origin Pro 9.0. The SEM image

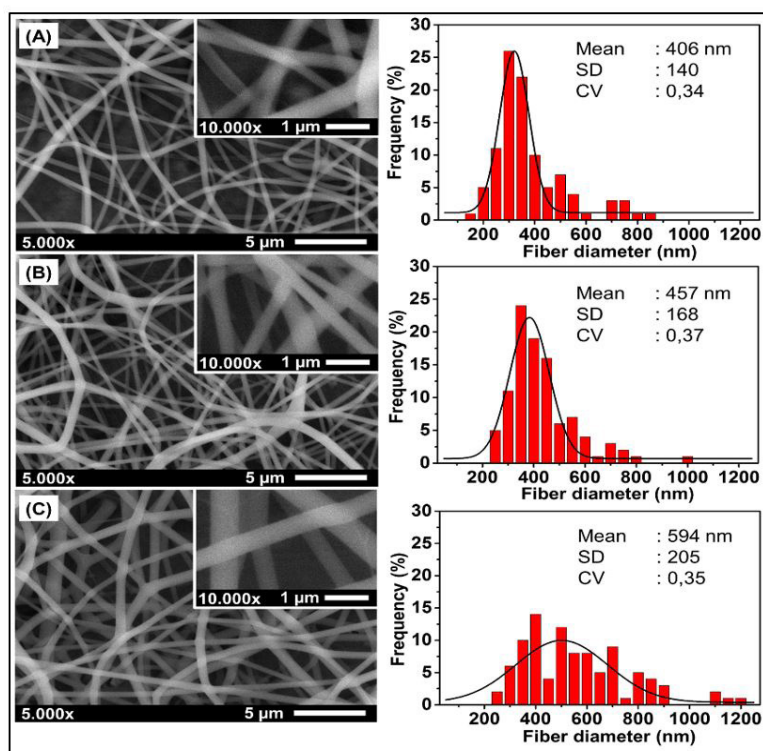


Figure 3. SEM image of 5000x and 10.000x magnification and nanoscale diameter distribution curve FA, FB, and FC with a various honey concentration (A) 0, (B) 1, and (C) 3% b/v

and the nanoscale diameter distribution curve of the electrospun fibers of FA, FB, and FC are shown in Figure 3.

Figure 3 showed that the morphology of the nanofibers produced were less uniform with the average diameters in the three formulas in the range of 100 nm to 1200 nm. The obtained nanofibers were porous and distributed randomly. From the data obtained the average diameter of nanofibers in FC (594 ± 205 nm) was the highest compared to FB (457 ± 168 nm) and FA (406 ± 140 nm). The solution FC had the highest compared to FB and FA so there was less stretch of polymer jet during the travel from the needle tip to the collector. As a result the diameter of the fibers was larger. Additionally, a solution with high viscosity can prevent the occurrence of secondary jets (jet splits), which usually can reduce the diameter of the nanofibers of causing the diameter to be non-uniform (Ramakrishna et al., 2005; Zhang et al., 2017).

The diameter of nanofibers with the range of 50-500 nm resembles natural extracellular matrices which can be utilized to accelerate homeostasis (Orue et al., 2017). Based on the data above, the diameter of the electrospun nanofibers for the three formulas were still above the permissible range. In addition, the diameter distribution of the nanofibers obtained was also not uniform in the three formulas based on the value of coefficient of

variation, or the ratio of standard deviation over mean value, which was above 0.3 (Matulevicius et al., 2015). There are several factors that cause highly diverse size of the fiber diameter. One of the contributing factors is high surface tension. In Table 2, it was shown the surface tension data of the three formulas were high, with the value of surface tension for FA, FB, and FC were respectively, 62.4; 61.8; 61.5 dyne/cm. Hence, the high surface tension caused the electrospinning process not to be able to run continuously. The high surface tension produces an unstable jet and there was a moment where the spherical solution droplets jumped from the tip of the needle directly to the collector. Therefore the electric voltage needs to be increased to overcome surface tension (Ramakrishna et al., 2005).

Characteristics of Blended Solution of PVA-GA-Honey with Addition of Triton X-100

In fact the electrospinning process did not run continuously and the diameter of the obtained nanofibers were very diverse. One way to overcome this is by adding nonionic surfactant namely Triton X-100. The purpose of adding surfactant was to reduce the surface tension of the solutions. Low surface tension is highly recommended because it will reduce the critical voltage required to produce jet of the solution from the Taylor cone (Chul et al., 2009). The nanofibers formed will be more smooth and uniform with the addition of surfactants (Ramakrishna et al., 2011; Risdian et al.,

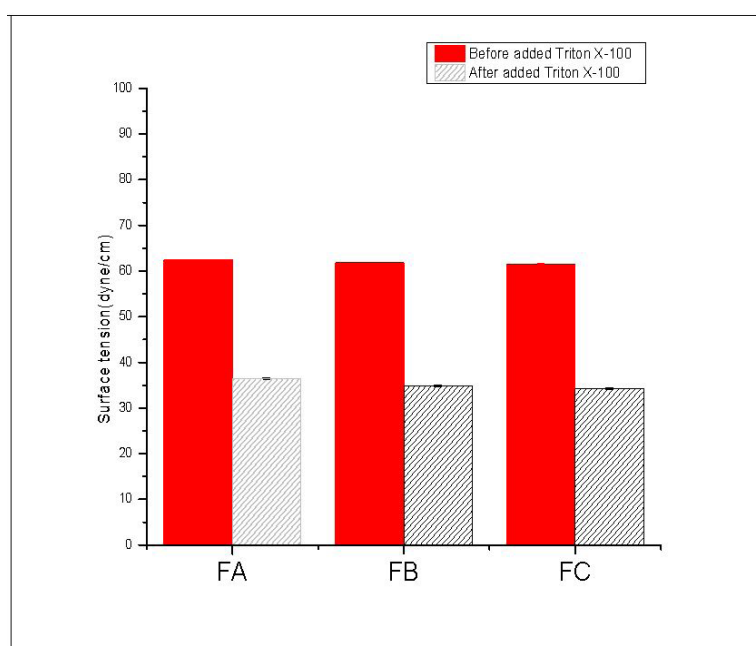


Figure 4. Surface tension of PVA-GA-Honey solution before and after addition of Triton X-100 (0,5 % v/v)

2015; Zhang et al., 2017).

PVA-GA and PVA-GA-honey solutions from the previous section was then added with Triton X-100 and then named as FAt, FBt, and FCt. The solutions were again characterized for its physical property of the solution, especially its surface tension. Based on the data in Figure 4 the surface tension in the FAt, FBt, and FCt formulas with the addition of Triton X-100 (0.05% v/v) decreased significantly compared to the FA, FB, and FC formulas which were respectively from 62, 61.8; 61.5 to 36.47; 34.83; 34.20 dyne/cm respectively. Similarly, Zhang et al (2017) have reported that the addition of 1% w/w Tween 80 into 25% w/v gelatin solution can significantly reduce the surface tension from 37.86 ± 0.04 to 19.76 ± 0.03 mN/m. The research conducted by Ziani et al (2011) reported that the addition of 40 mg Tween 20 into chitosan-solution polyethylene oxide can reduce surface tension significantly from 52.4 to 38.8 mN/m. The addition of surfactants to polymer solutions can modulate the interaction of polymers with solvents and polymers with surfactant through electrostatic, hydrophobic, and hydrogen bond interactions (Zhang et al., 2017).

Effects of Variation of Applied Voltage on Morphology and Diameter of FCt Nanofibers

Besides the solution parameters, the success in electrospinning is also determined by process parameters. Therefore, it is necessary to optimize the process parameters, including the applied voltage and flow rate. FCt was used in this optimization due to its highest

viscosity. After the most optimum process parameters had been obtained, the electrospun nanofibers were again prepared using the obtained optimized process parameters for all three solution formulas. The obtained nanofibers from the three formulas (FAt, FBt and FCt) possessed a smooth, uniform and beads-free morphology and nanoscale diameter.

The applied voltage used for the optimization were varied at 16, 18, and 20 kV, while other process parameters remained unchanged. The morphology of the nanofibers was characterized using SEM and the average diameter was calculated. The results of SEM image and the distribution diameter of nanofibers can be seen in Figure 5. It was shown that average diameters of the nanofibers reduced with increasing applied voltage. The average diameters of the nanofibers at a voltage 20 kV was smaller compared to nanofibers at 16 kV and 18 kV. The diameter size of the FCt nanofibers obtained with the applied voltages of 16; 18; and 20 kV were 333 ± 93 nm; 283 ± 57 nm; and 242 ± 39 nm respectively. At higher voltage, the higher electrical force causes larger stretching of the solution jet hence the diameter of the fibers becomes smaller. However, for applied voltage larger than 20 kV, beads occurred along the fibers since the very high voltage caused jump of jet so the time needed to stretch and vaporize the polymer chain entanglement in the solution was reduced (Ramakrishna et al., 2005). Meanwhile, at lower voltage, the diameter of the fibers increased since the amount of stretching was less. Based on these results, the optimum voltage that can be used to produce the best nanofibers was 18 kV.

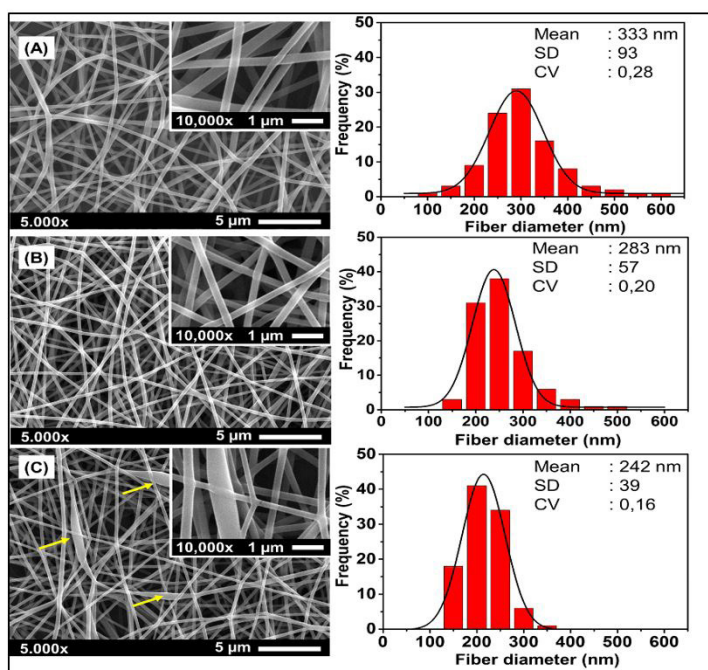


Figure 5. SEM image and the diameter distribution curve of FCt nanofibers with a various electrical voltage (A) 16, (B) 18, and (C) 20 kV magnification of 5000 x and 10.000x

Effect of Varying Flow rate on Morphology and Diameter of FCt Nanofibers

The flow rate of the electrospinning process was also optimized to obtain the best and uniform fibers. The flow rates varied in this optimization were 5, 8, and 10 $\mu\text{l}/\text{minutes}$, while other process parameters remained unchanged. The fibers were characterized using SEM to calculate the average diameter of each sample. The results of SEM image and the distribution of the diameter of fibers can be seen in Figure 6. Based on Figure 6, it can be shown that at higher flow rate caused larger diameter of fibers. The use of a flow rate at 5 $\mu\text{l}/\text{minutes}$ produced smooth and uniform nanofibers without beads. As the flow rate of the solution got larger, beads started to occur. The volume of the droplets at the tip of the needle increased at higher flow rate and caused the elongation process of the polymer chain became imperfect. Additionally, the increase of the flow rate causes the evaporation process did not fully dry up the solvent as the volume of jet is large. Hence the fibers arrived at the collector is still wet. (Ramakrishna et al., 2005).

Effect of Various Formulas with addition Triton X-100 on Morphology and Diameter of Nanofibers using Optimum Electric Voltage and Flow rate.

The nanofibers morphology of the three formulas were observed using a scanning electron microscope (SEM). The results of this morphological observation is shown in Figure 7. The addition of Triton x-100 into all three FAt, FBt, and FCt formulas improved the morphology

and diameter size of nano fibers produced compared to the previous formula without Triton X-100. The morphology of obtained electrospun nanofibers was smooth, randomly oriented and uniform as indicated by the coefficient of variation to be below 0.3. Based on the results of research by Risdian et al (2015), the presence of Triton X-100 at concentration 0.05% v/v can affect the morphology of the nanofibers produced. The addition of a small amount of non-ionic surfactant to the polymer solution decreases the surface tension of the solution so that it can increase the electrospinnability of the solution and increase its reproducibility during the electrospinning process (Ziani et al., 2011). A similar study also stated that the addition of non-ionic surfactants (Tween 20) to chitosan-polyethylene oxide solutions could improve the ability of the solution to be electrically spun. The nanofibers has better morphology with added surfactant compared to the fibers produced from blended solution without the addition of Tween 20 (Ziani et al., 2011). According to Bertonecelj. V et al (2014), the morphology and size of the diameter of nanofibers affect the cell proliferation process. The process of cell proliferation *invitro* decreases if there is a defect/bead in nanofibers. The diameter of nanofibers at range 50-500nm resembles natural extracellular matrices which can accelerate the process of homeostasis in wounds (Orue et al., 2017). From the data obtained the average diameter of nanofibers produced from FCt, FBt, and FAt was 283 ± 57 nm; 266 ± 45 nm; and 244 ± 45 nm, respectively. The large diameter of FCt was due to its highest viscosity. The nanofibers diameter increased with increasing viscosity of solution (Ramakrishna et al.,

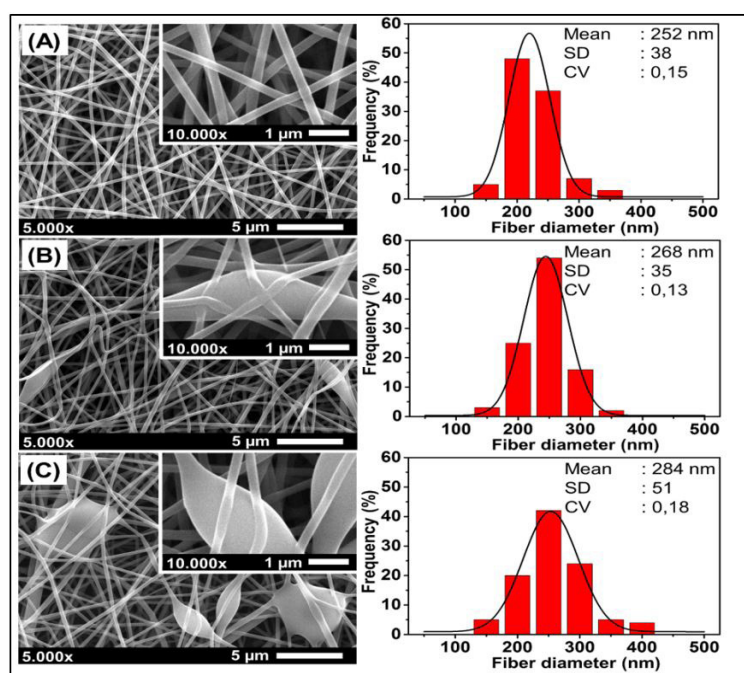


Figure 6 . SEM image and the diameter distribution curve of FCt nanofibers with a various flow rate (A) 5, (B) 8, and (C) 10 $\mu\text{l}/\text{min}$ magnification of 5000 x and 10.000 x

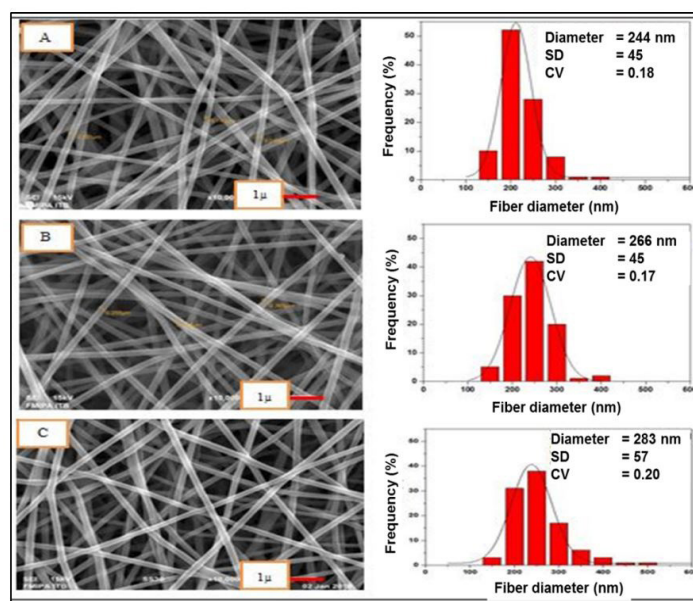


Figure 7. SEM image and the diameter distribution curve of nanofibers with a various formulas (A) FAt, (B) FBt, and (C) FCt magnification of 10.000 x

2005). From the data, it is shown that the morphology and diameter size of the obtained nanofibers were within the range of the size of the natural extracellular matrix.

CONCLUSION

PVA-GA-honey-based nanofibers have been successfully optimized with good morphology, smooth surface and uniform diameter in nanoscale which is mimics the

natural extracellular matrix. The most optimum process parameters for produce fibers was the applied voltage and flow rate at 18 KV and 5 $\mu\text{l}/\text{min}$. The result showed that FCt (with 3% of honey and 0.05% of Triton X-100) was the best formula with the average fibers diameter of FCt obtained 283 ± 57 nm that still within the range of the extra cellular matrix. Additionally, at higher honey content in FCt, the wound healing process could accelerated and protected the stability of growth

factors during the electrospinning process. Furthermore FCt have the potential to be further developed as carrier for peptide and protein drugs.

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